(FILE 'HOME' ENTERED AT 16:31:31 ON 26 JAN 2002) FILE 'CAPLUS' ENTERED AT 16:31:46 ON 26 JAN 2002 357 S SUBMERG? COMBUST? L1L2 1 S CULLET (3N) ORGANIC FILE 'STNGUIDE' ENTERED AT 16:33:58 ON 26 JAN 2002 FILE 'CAPLUS' ENTERED AT 16:34:07 ON 26 JAN 2002 1 S (CULLET (5A) ORGANIC) L3 1 S (CULLET (S) ORGANIC) L4L5 5 S (CULLET AND ORGANIC) 590982 S (CULLET OR GLASS) L6590982 S (CULLET OR GLASS) L7 L8 587670 S (WASTE OR ORGANIC) L10 ANSWER 1 OF 3 CAPLUS COPYRIGHT 2002 ACS AN 1998:174931 CAPLUS 128:234216 DN Manufacture of supplementary cementitious materials from industrial ΤI Mishulovich, Alex; Bhatty, Javed I.; Abbasi, Hamid A.; Rue, David; Olabin, ΑU Vladimir M.; Pioro, Leonard S. Construction Technology Laboratories, Skokie, IL, USA CS Proc. Int. Conf. Incineration Therm. Treat. Technol. (1996), 297-301 Publisher: University of California, Irvine, Irvine, Calif. CODEN: 65TTAP DTConference LA English 58-1 (Cement, Concrete, and Related Building Materials) CC Section cross-reference(s): 60 Supplementary cementitious materials (SCM) were manufd. by melting and AB vitrification of specially designed blends of wastes with the addn. of inexpensive natural materials (limestone, sand, shale, etc.). This approach opens an outlet for the rational use of wastes and reduces carbon dioxide emission usually assocd. with prodn. of conventional portland cement. The paper summarizes the results of the bench top phase of formulation and testing of SCMs prepd. from Illinois coal ash with the addn. of inexpensive natural or waste materials, such as limestone or cement kiln dust. Selection of the prospective compns. was based on the anal. of phase equil. in the system CaO-SiO2-Al2O3. Compns. were chosen that melt at temps. <1250.degree.. These compns. were realized by mixing the ingredients in the calcd. proportions. Performance of the produced materials was tested in blended cements and concretes. Blended cements incorporating SCMs are not only competitive in terms of strength but have an addnl. advantage of preventing or greatly reducing deleterious chem. reactions between the cement paste and concrete aggregates. This improves the durability of concrete. Unlike conventional portland cements, the supplementary cementing materials should be produced in the form of glasses to provide the necessary chem. reactivity of the product. Besides, vitrification prevents leaching of the trace elements present in the source materials. Submerged gas combustion was suggested and tested as the process of choice for commercialization of this technol. Pilot testing of the submerged combustion melter begins this year in a 250-kg/h test facility. Further studies are under way to finalize the prodn. process parameters and to investigate the products performance. supplementary cementitious material industrial waste STΙT Wastes (industrial; manuf. of supplementary cementitious materials from industrial wastes) ΙT Cement (construction material)

Recycling (manuf. of supplementary cementitious materials from industrial wastes) L10 ANSWER 2 OF 3 CAPLUS COPYRIGHT 2002 ACS 1994:282461 CAPLUS AN DN 120:282461 Development of spent solvent treatment process by a submerged ΤI combustion technique Uchiyama, Gunzo; Maeda, Mitsuru; Fujine, Sachio; Amakawa, Masayuki; ΑU Uchida, Katsuhide; Chida, Mitsuhisa Dep. Fuel Cycle Saf. Res., Japan At. Energy Res. Inst., Tokai, 319-11, CS J. Nucl. Sci. Technol. (1994), 31(3), 228-39 SO CODEN: JNSTAX; ISSN: 0022-3131 Journal DTEnglish LA 71-5 (Nuclear Technology) CC An exptl. study using a bench-scale equipment of 1 kg-stimulated spent AΒ solvents per h was conducted in order to evaluate the applicability of a submerged combustion technique for the treatment of spent solvents contaminated with TRU elements. This report describes the exptl. results on the combustion characteristics of the simulated spent solvents of tri-Bu phosphate and/or n-dodecane, and on the distribution behaviors of combustion products such as phosphoric acid, Ru, I, Zr and lanthanides as TRU simulants in the submerged combustion process. Also the exptl. results of TRU sepn. from phosphoric acid soln. by copptn. using bismuth phosphate are reported. The submerged combustion technique was applicable to the treatment of spent solvents including the distn. residues of the solvent. Based on the exptl. data, a new treatment process of spent solvent was proposed which consisted of submerged combustion, co-pptn. using bismuth phosphate, ceramic membrane filtration, cementation of TRU lean phosphate, and vitrification of TRU rich waste. submerged combustion spent fuel reprocessing solvent; STtributyl phosphate submerged combustion; dodecane submerged combustion; lanthanide transuranium simulant submerged combustion; copptn phosphate transuranium sepn radioactive waste; cementation phosphate sepn radioactive waste; filtration spent solvent fuel reprocessing; vitrification transuranium sepn fuel reprocessing waste ΙT Gamma ray (-emitters, recovery of, in treatment process for spent fuel reprocessing solvents) Rare earth metals, reactions IT RL: RCT (Reactant) (combustion of transuranic simulant, in spent fuel reprocessing solvent treatment process) Hydrocarbons, preparation IT RL: FORM (Formation, nonpreparative) (formation of, in simulated spent fuel reprocessing solvent treatment by submerged combustion) ITEvaporation Neutralization (in nuclear fuel reprocessing spent solvent treatment process) TT (of transuranium-element depleted phosphate wastes, in treatment process for spent fuel reprocessing solvents) Transuranium elements ΙT RL: PROC (Process) (recovery of, in treatment process for spent fuel reprocessing solvents) ΤТ Radioactive wastes

(spent solvent treatment process for fuel reprocessing, submerged combustion technique in) Nuclear reactor fuel reprocessing ΙT (spent solvent treatment process, submerged combustion technique in) Filters and Filtering materials IT (ceramic, in nuclear fuel reprocessing spent solvent treatment process) TΤ (co-, in nuclear fuel reprocessing spent solvent treatment process) Glass, oxide IT RL: PROC (Process) (radioactive-waste, transuranium-element enriched, in treatment process for spent fuel reprocessing solvents) ΙT Combustion (submerged, in nuclear fuel reprocessing spent solvent treatment process) 12587-47-2P, Beta particle ΙT RL: PREP (Preparation) (-emitters, recovery of, in submerged combustion treatment process for spent fuel reprocessing solvents) IT 34513-98-9 RL: PROC (Process) (combustion of simulated spent fuel reprocessing solvent contg., radioactive waste disposal in relation to) 74-88-4, Methyl iodide, reactions 75-03-6, Ethyl iodide ΙT Propyl iodide RL: RCT (Reactant) (combustion of simulated spent fuel reprocessing solvent contg., radioactive waste disposal in relation to) 10138-01-9, Europium trinitrate 10108-73-3, Cerium trinitrate IT 13826-66-9, Zirconyl dinitrate 10361-83-8, Samarium trinitrate RL: PROC (Process) (combustion of transuranic simulant, in spent fuel reprocessing solvent treatment process) 126-73-8, Tributyl phosphate, reactions 112-40-3, Dodecane IT RL: RCT (Reactant) (combustion of, in simulated spent fuel reprocessing solvent treatment process, radioactive waste issues in relation to) IΤ 7782-44-7 RL: PROC (Process) (combustion, submerged, in nuclear fuel reprocessing spent solvent treatment process) 13847-18-2, Barium phosphate IT RL: PROC (Process) (copptn. with, of transuranics in simulated spent fuel reprocessing solvent treatment process) 630-08-0P, Carbon monoxide, 124-38-9P, Carbon dioxide, preparation ΙT 7782-44-7P, Oxygen, preparation preparation RL: FORM (Formation, nonpreparative); PREP (Preparation) (formation of, in simulated spent fuel reprocessing solvent treatment by submerged combustion) 15454-31-6P, Iodate (IO3-) 7553-56-2P, Iodine, preparation IT 20461-54-5P, Iodide, preparation RL: PREP (Preparation) (recovery of, from combustion of simulated spent fuel reprocessing solvent contg. alkyl iodides) 7440-18-8P, Ruthenium, preparation IT RL: PREP (Preparation) (recovery of, from combustion of simulated spent fuel reprocessing solvent contg. ruthenium nitrosyl nitrate) 7440-18-8DP, Ruthenium, phosphate complexes IT RL: PREP (Preparation) (recovery of, from combustion of simulated spent fuel reprocessing

solvent contg. ruthenium nitrosyl nitrate, radioactive waste disposal in relation to) 13454-71-2P, Cerium monophosphate ΙT RL: PREP (Preparation) (recovery of, from combustion of spent TBP solvent contg. cerium nitrate, fuel reprocessing waste issues in relation to) 13537-10-5P, Europium(III) phosphate IT RL: PREP (Preparation) (recovery of, from combustion of spent TBP solvent contg. europium nitrate, fuel reprocessing waste issues in relation to) 13465-57-1P, Samarium(III) phosphate TΤ RL: PREP (Preparation) (recovery of, from combustion of spent TBP solvent contg. samarium nitrate, fuel reprocessing waste issues in relation to) 13565-97-4P, Zirconium pyrophosphate ΙT RL: PREP (Preparation) (recovery of, from combustion of spent TBP solvent contg. zirconyl nitrate, fuel reprocessing waste issues in relation to) 7440-07-5P, Plutonium, preparation IT RL: PREP (Preparation) (recovery of, from phosphoric acid solns. by copptn. with barium phosphate, spent fuel reprocessing solvent waste in relation to) 7664-38-2P, Phosphoric acid, preparation ΤT RL: PREP (Preparation) (recovery of, from tri-Bu phosphate combustion in simulated spent fuel reprocessing solvent treatment process) L10 ANSWER 3 OF 3 CAPLUS COPYRIGHT 2002 ACS 1964:467389 CAPLUS AN 61:67389 DN OREF 61:11643c-d Recovery of sulfuric acid and metal salt from spent sulfuric acid. IX. TΤ Treatment of spent sulfuric acid liquor by middle scale apparatus. 2. Concentration of spent sulfuric acid liquor by the submerged combustion apparatus Kubo, Teruichiro; Taniguchi, Masao; Senda, Shokichi; Sato, Katsuaki ΑU Tokyo Inst. Technol CS Kogyo Kagaku Zasshi (1964), 67, 869 SO DT Journal Unavailable LA 17 (Industrial Inorganic Chemicals) CC By use of a submerged combustion vessel having a AB capacity of 400 1. the concn. of dil. H2SO4 and waste H2SO4 was examd. The inner face of the steel vessel was lined with glass. The burner for the combustion of natural gas was made of graphite in dil. acid and of steel in concd. acid. Heat efficiency was 78.5-85.8% at concns. of 10.9-42.5% and 61.2-78.8% at concns. of 42.5-85.1%. In the case of xwaste acid concn. almost the same results were obtained. (FILE 'HOME' ENTERED AT 16:31:31 ON 26 JAN 2002) FILE 'CAPLUS' ENTERED AT 16:31:46 ON 26 JAN 2002 357 SEA PLU=ON SUBMERG? COMBUST? L10 S CULLET NEAR3 ORGANIC L*** DEL 1 SEA PLU=ON CULLET (3A) ORGANIC L2 FILE 'STNGUIDE' ENTERED AT 16:33:58 ON 26 JAN 2002 SET LINE 250 SET DETAIL OFF

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    combustion process. I
    Osada, Takeshi; Miyauchi, Yoshio
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    Toray Eng. Co., Ltd., Japan
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    Treatment of organic waste liquor by the submerged
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    combustion process
    Miyauchi, Yoshio; Osada, Takeshi
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    Toray Eng. Co., Ltd., Osaka, Japan
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     PPM (1978), 9(5), 25-35
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L12 ANSWER 4 OF 6 CAPLUS COPYRIGHT 2002 ACS
     1975:174055 CAPLUS
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     Recovery of Group VIII metals from solutions of rare metal-
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     organophosphorus complexes in organic solvents
     Onoda, Takeru; Tsunoda, Yoshitoshi; Nomura, Takao; Nonaka, Takehisa;
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     Kurashiki, Okayama; Masuyama, Tetsuo
    Mitsubishi Chemical Industries Co., Ltd., Japan
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     Ger. Offen., 15 pp.
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L12 ANSWER 5 OF 6 CAPLUS COPYRIGHT 2002 ACS
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     Waste treatment process based on the submerged
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     combustion technology
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     Tsuruta, Hidemasa
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     Nittetsu Chem. Eng. Ltd., Japan
     Sekiyu Gakkai Shi (1973), 16(8), 646-50
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     CODEN: SKGSAE
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Journal DΤ LА Japanese L12 ANSWER 6 OF 6 CAPLUS COPYRIGHT 2002 ACS 1971:467222 CAPLUS 75:67222 DN Recovering hydrogen chloride from a spent organochlorine compound ΤI Ezaki, Shigeho IN Yawata Chemical Engineering Co., Ltd. PA SO U.S., 5 pp. CODEN: USXXAM DTPatent English LΑ FAN.CNT 1 APPLICATION NO. DATE KIND DATE PATENT NO. ____ _____ US 1969-811905 19690401 US 3589864 A 19710629 PRAI JP 1968-21404 19680403 ANSWER 1 OF 2 CAPLUS COPYRIGHT 2002 ACS L14 1998:174931 CAPLUS AN 128:234216 DN Manufacture of supplementary cementitious materials from industrial ΤI Mishulovich, Alex; Bhatty, Javed I.; Abbasi, Hamid A.; Rue, David; Olabin, ΑU Vladimir M.; Pioro, Leonard S. Construction Technology Laboratories, Skokie, IL, USA CS

- Proc. Int. Conf. Incineration Therm. Treat. Technol. (1996), 297-301 Publisher: University of California, Irvine, Irvine, Calif. CODEN: 65TTAP
- DT Conference
- LA English
- CC 58-1 (Cement, Concrete, and Related Building Materials) Section cross-reference(s): 60
- Supplementary cementitious materials (SCM) were manufd. by melting and AΒ vitrification of specially designed blends of wastes with the addn. of inexpensive natural materials (limestone, sand, shale, etc.). This approach opens an outlet for the rational use of wastes and reduces carbon dioxide emission usually assocd. with prodn. of conventional portland cement. The paper summarizes the results of the bench top phase of formulation and testing of SCMs prepd. from Illinois coal ash with the addn. of inexpensive natural or waste materials, such as limestone or cement kiln dust. Selection of the prospective compns. was based on the anal. of phase equil. in the system CaO-SiO2-Al2O3. Compns. were chosen that melt at temps. <1250.degree.. These compns. were realized by mixing the ingredients in the calcd. proportions. Performance of the produced materials was tested in blended cements and concretes. Blended cements incorporating SCMs are not only competitive in terms of strength but have an addnl. advantage of preventing or greatly reducing deleterious chem. reactions between the cement paste and concrete aggregates. This improves the durability of concrete. Unlike conventional portland cements, the supplementary cementing materials should be produced in the form of glasses to provide the necessary chem. reactivity of the product. Besides, vitrification prevents leaching of the trace elements present in the source materials. Submerged gas combustion was suggested and tested as the process of choice for commercialization of this technol. Pilot testing of the submerged combustion melter begins this year in a 250-kg/h test facility. Further studies are under way to finalize the prodn. process parameters and to investigate the products performance.

ST supplementary cementitious material industrial waste

ΙT Wastes (industrial; manuf. of supplementary cementitious materials from industrial wastes) Cement (construction material) ΙT Recycling (manuf. of supplementary cementitious materials from industrial wastes) L14 ANSWER 2 OF 2 CAPLUS COPYRIGHT 2002 ACS 1994:282461 CAPLUS ΑN 120:282461 DN Development of spent solvent treatment process by a submerged ΤI combustion technique Uchiyama, Gunzo; Maeda, Mitsuru; Fujine, Sachio; Amakawa, Masayuki; ΑU Uchida, Katsuhide; Chida, Mitsuhisa Dep. Fuel Cycle Saf. Res., Japan At. Energy Res. Inst., Tokai, 319-11, CS Japan J. Nucl. Sci. Technol. (1994), 31(3), 228-39 SO CODEN: JNSTAX; ISSN: 0022-3131 DTJournal English LΑ 71-5 (Nuclear Technology) CC An exptl. study using a bench-scale equipment of 1 kg-stimulated spent AΒ solvents per h was conducted in order to evaluate the applicability of a submerged combustion technique for the treatment of spent solvents contaminated with TRU elements. This report describes the exptl. results on the combustion characteristics of the simulated spent solvents of tri-Bu phosphate and/or n-dodecane, and on the distribution behaviors of combustion products such as phosphoric acid, Ru, I, Zr and lanthanides as TRU simulants in the submerged combustion process. Also the exptl. results of TRU sepn. from phosphoric acid soln. by copptn. using bismuth phosphate are reported. The submerged combustion technique was applicable to the treatment of spent solvents including the distn. residues of the solvent. Based on the exptl. data, a new treatment process of spent solvent was proposed which consisted of submerged combustion, co-pptn. using bismuth phosphate, ceramic membrane filtration, cementation of TRU lean phosphate, and vitrification of TRU rich waste. submerged combustion spent fuel reprocessing solvent; ST tributyl phosphate submerged combustion; dodecane submerged combustion; lanthanide transuranium simulant submerged combustion; copptn phosphate transuranium sepn radioactive waste; cementation phosphate sepn radioactive waste; filtration spent solvent fuel reprocessing; vitrification transuranium sepn fuel reprocessing waste ΙT Gamma ray (-emitters, recovery of, in treatment process for spent fuel reprocessing solvents) Rare earth metals, reactions RL: RCT (Reactant) (combustion of transuranic simulant, in spent fuel reprocessing solvent treatment process) Hydrocarbons, preparation ΙT RL: FORM (Formation, nonpreparative) (formation of, in simulated spent fuel reprocessing solvent treatment by submerged combustion) Evaporation IT Neutralization (in nuclear fuel reprocessing spent solvent treatment process) ΙT (of transuranium-element depleted phosphate wastes, in treatment process for spent fuel reprocessing solvents) Transuranium elements IT

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RL: PROC (Process)
        (recovery of, in treatment process for spent fuel reprocessing
     Radioactive wastes
IT
        (spent solvent treatment process for fuel reprocessing,
        submerged combustion technique in)
     Nuclear reactor fuel reprocessing
ΙT
        (spent solvent treatment process, submerged
        combustion technique in)
     Filters and Filtering materials
ΙT
        (ceramic, in nuclear fuel reprocessing spent solvent treatment process)
     Precipitation
ΙT
        (co-, in nuclear fuel reprocessing spent solvent treatment process)
     Glass, oxide
IT
     RL: PROC (Process)
        (radioactive-waste, transuranium-element enriched, in
        treatment process for spent fuel reprocessing solvents)
     Combustion
IT
        (submerged, in nuclear fuel reprocessing spent solvent treatment
        process)
     12587-47-2P, Beta particle
IT
     RL: PREP (Preparation)
        (-emitters, recovery of, in submerged combustion
        treatment process for spent fuel reprocessing solvents)
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     34513-98-9
     RL: PROC (Process)
        (combustion of simulated spent fuel reprocessing solvent contg.,
        radioactive waste disposal in relation to)
                                         75-03-6, Ethyl iodide
                                                                  107-08-4,
     74-88-4, Methyl iodide, reactions
IT
     Propyl iodide
     RL: RCT (Reactant)
        (combustion of simulated spent fuel reprocessing solvent contg.,
        radioactive waste disposal in relation to)
                                     10138-01-9, Europium trinitrate
     10108-73-3, Cerium trinitrate
ΙT
                                       13826-66-9, Zirconyl dinitrate
     10361-83-8, Samarium trinitrate
     RL: PROC (Process)
        (combustion of transuranic simulant, in spent fuel reprocessing solvent
        treatment process)
                           126-73-8, Tributyl phosphate, reactions
     112-40-3, Dodecane
IT
     RL: RCT (Reactant)
        (combustion of, in simulated spent fuel reprocessing solvent treatment
        process, radioactive waste issues in relation to)
     7782-44-7
ΙT
     RL: PROC (Process)
         (combustion, submerged, in nuclear fuel reprocessing spent solvent
         treatment process)
     13847-18-2, Barium phosphate
ΙT
     RL: PROC (Process)
         (copptn. with, of transuranics in simulated spent fuel reprocessing
         solvent treatment process)
                                               630-08-0P, Carbon monoxide,
      124-38-9P, Carbon dioxide, preparation
ΙT
                  7782-44-7P, Oxygen, preparation
     preparation
     RL: FORM (Formation, nonpreparative); PREP (Preparation)
         (formation of, in simulated spent fuel reprocessing solvent treatment
         by submerged combustion)
                                        15454-31-6P, Iodate (IO3-)
      7553-56-2P, Iodine, preparation
 IT
      20461-54-5P, Iodide, preparation
      RL: PREP (Preparation)
         (recovery of, from combustion of simulated spent fuel reprocessing
         solvent contg. alkyl iodides)
      7440-18-8P, Ruthenium, preparation
 IT
      RL: PREP (Preparation)
         (recovery of, from combustion of simulated spent fuel reprocessing
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solvent contg. ruthenium nitrosyl nitrate)
    7440-18-8DP, Ruthenium, phosphate complexes
IT
     RL: PREP (Preparation)
        (recovery of, from combustion of simulated spent fuel reprocessing
        solvent contg. ruthenium nitrosyl nitrate, radioactive waste
        disposal in relation to)
     13454-71-2P, Cerium monophosphate
TΤ
     RL: PREP (Preparation)
        (recovery of, from combustion of spent TBP solvent contg. cerium
        nitrate, fuel reprocessing waste issues in relation to)
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ΙT
     RL: PREP (Preparation)
        (recovery of, from combustion of spent TBP solvent contg. europium
        nitrate, fuel reprocessing waste issues in relation to)
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IT
     RL: PREP (Preparation)
        (recovery of, from combustion of spent TBP solvent contg. samarium
        nitrate, fuel reprocessing waste issues in relation to)
     13565-97-4P, Zirconium pyrophosphate
IT
     RL: PREP (Preparation)
        (recovery of, from combustion of spent TBP solvent contg. zirconyl
        nitrate, fuel reprocessing waste issues in relation to)
     7440-07-5P, Plutonium, preparation
IT
     RL: PREP (Preparation)
        (recovery of, from phosphoric acid solns. by copptn. with barium
        phosphate, spent fuel reprocessing solvent waste in relation
     7664-38-2P, Phosphoric acid, preparation
ΙT
     RL: PREP (Preparation)
        (recovery of, from tri-Bu phosphate combustion in simulated spent fuel
        reprocessing solvent treatment process)
     (FILE 'HOME' ENTERED AT 16:31:31 ON 26 JAN 2002)
     FILE 'CAPLUS' ENTERED AT 16:31:46 ON 26 JAN 2002
            357 SEA PLU=ON SUBMERG? COMBUST?
L1
              0 S CULLET NEAR3 ORGANIC
L*** DEL
              1 SEA PLU=ON CULLET (3A) ORGANIC
L2
     FILE 'STNGUIDE' ENTERED AT 16:33:58 ON 26 JAN 2002
                SET LINE 250
                SET DETAIL OFF
     FILE 'CAPLUS' ENTERED AT 16:34:07 ON 26 JAN 2002
                SET LINE LOGIN
                SET DETAIL LOGIN
                SET NOTICE 1000 SEARCH
              1 SEA PLU=ON (CULLET (5A) ORGANIC)
L3
                            (CULLET (S) ORGANIC)
              1 SEA PLU=ON
L4
              5 SEA PLU=ON
                            (CULLET AND ORGANIC)
L5
                            (CULLET OR GLASS)
         590982 SEA PLU=ON
Lб
                            (CULLET OR GLASS)
         590982 SEA PLU=ON
L7
                            (WASTE OR ORGANIC)
         587670 SEA PLU=ON
\Gamma8
                 SET NOTICE LOGIN SEARCH
          20742 SEA PLU=ON L6 AND L8
1.9
              3 SEA PLU=ON L1 AND L9
L10
                 D 1-3 ALL
                 D COST FULL
              6 SEA PLU=ON L1 AND ORGANIC
L11
               6 SEA PLU=ON L11 NOT L10
L12
                 D 1-6
           2233 SEA PLU=ON VITRIFICATION AND WASTE
```

L13

FILE HOME

FILE CAPLUS

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FILE COVERS 1907 - 26 Jan 2002 VOL 136 ISS 5 FILE LAST UPDATED: 25 Jan 2002 (20020125/ED)

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FILE STNGUIDE FILE CONTAINS CURRENT INFORMATION. LAST RELOADED: Jan 25, 2002 (20020125/UP).

FILE & COST CENTER	YTITMAUQ	@	RATE	ESTIMATED COST U.S. DOLLARS
HOME FILE COST=				
CONNECT HOURS	0.01	@	15.00	0.15
INTERNET HOURS	0.01	9	6.00	0.06
CAPLUS FILE COST=				
CONNECT HOURS	0.04	@	32.00	1.28
INTERNET HOURS	0.04	@	6.00	0.24
SEARCH TERMS IN FIELD BI	7	@	1.55	10.85
CAPLUS FEE (5%)	12.13	9	0.05	0.61
STNGUIDE FILE COST=				
SFE CONNECT HOURS	0.01	@	0.00	0.00
INTERNET HOURS	0.01	9	6.00	0.06
CAPLUS FILE COST=				
CONNECT HOURS	0.09	9	32.00	2.88
INTERNET HOURS	0.09	9	6.00	0.54
DISPLAYS IN FORMAT ABS	5	9	1.27	6.35
DISPLAYS IN FORMAT BIB	11	@	0.91	10.01

DISPLAYS IN FORMAT IND SEARCH TERMS IN FIELD BI CAPLUS FEE (5%)	5 15 43.84	@ 0.2 @ 1.5 ! @ 0.0	55	1.35 23.25 2.19
SUMMARY BY FILE AND	cos	CENTER	HOURS	ESTIMATED COST U.S. DOLLARS
HOME FILE CAPLUS FILE STNGUIDE FILE		(NONE) (NONE) (NONE)	0.13	0.21
COSTS INCLUDE TELECOMMUNICATIO	N FEES		0.15	0.90
SUMMARY BY	COST CENTER		HOURS	ESTIMATED COST
YOUR TOTAL SESSION COSTS ARE	(NONE)		0.15 0.15	59.82
DISCOUNT AMOUNTS (FOR QUALIFYI	NG ACCOUNTS)	SIN	NCE FILE	
CA SUBSCRIBER PRICE			-3.10	-3.10

IN FILE 'CAPLUS' AT 16:49:41 ON 26 JAN 2002

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ANSWER 2 OF 2 CAPLUS COPYRIGHT 2002 ACS
     1994:282461 CAPLUS
AN
     120:282461
DN
TT
     Development of spent solvent treatment process by a submerged
     combustion technique
ΑU
     Uchiyama, Gunzo; Maeda, Mitsuru; Fujine, Sachio; Amakawa, Masayuki;
     Uchida, Katsuhide; Chida, Mitsuhisa
     Dep. Fuel Cycle Saf. Res., Japan At. Energy Res. Inst., Tokai, 319-11,
CS
     Japan
SO
     J. Nucl. Sci. Technol. (1994), 31(3), 228-39
     CODEN: JNSTAX; ISSN: 0022-3131
DT
     Journal
LΑ
     English
CC
     71-5 (Nuclear Technology)
AB
     An exptl. study using a bench-scale equipment of 1 kg-stimulated spent
     solvents per h was conducted in order to evaluate the applicability of a
     submerged combustion technique for the treatment of
     spent solvents contaminated with TRU elements. This report describes the
     exptl. results on the combustion characteristics of the simulated spent
     solvents of tri-Bu phosphate and/or n-dodecane, and on the distribution
     behaviors of combustion products such as phosphoric acid, Ru, I, Zr and
     lanthanides as TRU simulants in the submerged combustion
     process. Also the exptl. results of TRU sepn. from phosphoric acid soln.
     by copptn. using bismuth phosphate are reported. The submerged
     combustion technique was applicable to the treatment of spent
     solvents including the distn. residues of the solvent. Based on the
     exptl. data, a new treatment process of spent solvent was proposed which
     consisted of submerged combustion, co-pptn. using
     bismuth phosphate, ceramic membrane filtration, cementation of TRU lean
     phosphate, and vitrification of TRU rich waste.
     submerged combustion spent fuel reprocessing solvent;
     tributyl phosphate submerged combustion; dodecane
     submerged combustion; lanthanide transuranium simulant
     submerged combustion; copptn phosphate transuranium sepn
     radioactive waste; cementation phosphate sepn radioactive
     waste; filtration spent solvent fuel reprocessing;
     vitrification transuranium sepn fuel reprocessing waste
IT
     Gamma ray
        (-emitters, recovery of, in treatment process for spent fuel
        reprocessing solvents)
     Rare earth metals, reactions
ΙT
     RL: RCT (Reactant)
        (combustion of transuranic simulant, in spent fuel reprocessing solvent
        treatment process)
ΙT
     Hydrocarbons, preparation
     RL: FORM (Formation, nonpreparative)
        (formation of, in simulated spent fuel reprocessing solvent treatment
        by submerged combustion)
IT
     Evaporation
     Neutralization
        (in nuclear fuel reprocessing spent solvent treatment process)
IT
     Cementation
        (of transuranium-element depleted phosphate wastes, in
        treatment process for spent fuel reprocessing solvents)
IT
    Transuranium elements
    RL: PROC (Process)
        (recovery of, in treatment process for spent fuel reprocessing
        solvents)
IT
    Radioactive wastes
        (spent solvent treatment process for fuel reprocessing,
        submerged combustion technique in)
IT
    Nuclear reactor fuel reprocessing
```

L14

```
(spent solvent treatment process, submerged
        combustion technique in)
     Filters and Filtering materials (ceramic, in nuclear fuel reprocessing spent solvent treatment process)
IT
     Precipitation
\mathbf{IT}
        (co-, in nuclear fuel reprocessing spent solvent treatment process)
     Glass, oxide
IT
     RL: PROC (Process)
        (radioactive-waste, transuranium-element enriched, in
        treatment process for spent fuel reprocessing solvents)
IT
        (submerged, in nuclear fuel reprocessing spent solvent treatment
        process)
     12587-47-2P, Beta particle
IT
     RL: PREP (Preparation)
        (-emitters, recovery of, in submerged combustion
        treatment process for spent fuel reprocessing solvents)
ΙT
     34513-98-9
     RL: PROC (Process)
        (combustion of simulated spent fuel reprocessing solvent contg.,
        radioactive waste disposal in relation to)
IT
     74-88-4, Methyl iodide, reactions
                                          75-03-6, Ethyl iodide
     Propyl iodide
     RL: RCT (Reactant)
        (combustion of simulated spent fuel reprocessing solvent contg.,
        radioactive waste disposal in relation to)
     10108-73-3, Cerium trinitrate
                                      10138-01-9, Europium trinitrate
IT
     10361-83-8, Samarium trinitrate 13826-66-9, Zirconyl dinitrate
     RL: PROC (Process)
        (combustion of transuranic simulant, in spent fuel reprocessing solvent
        treatment process)
                          126-73-8, Tributyl phosphate, reactions
IT
     112-40-3, Dodecane
     RL: RCT (Reactant)
        (combustion of, in simulated spent fuel reprocessing solvent treatment
        process, radioactive waste issues in relation to)
IT
     7782-44-7
     RL: PROC (Process)
        (combustion, submerged, in nuclear fuel reprocessing spent solvent
        treatment process)
IT
     13847-18-2, Barium phosphate
     RL: PROC (Process)
        (copptn. with, of transuranics in simulated spent fuel reprocessing
        solvent treatment process)
IT
     124-38-9P, Carbon dioxide, preparation
                                               630-08-0P, Carbon monoxide,
     preparation
                  7782-44-7P, Oxygen, preparation
     RL: FORM (Formation, nonpreparative); PREP (Preparation)
        (formation of, in simulated spent fuel reprocessing solvent treatment
        by submerged combustion)
IT
     7553-56-2P, Iodine, preparation
                                        15454-31-6P, Iodate (IO3-)
     20461-54-5P, Iodide, preparation
     RL: PREP (Preparation)
        (recovery of, from combustion of simulated spent fuel reprocessing
        solvent contg. alkyl iodides)
ŤΨ
     7440-18-8P, Ruthenium, preparation
     RL: PREP (Preparation)
        (recovery of, from combustion of simulated spent fuel reprocessing
        solvent contq. ruthenium nitrosyl nitrate)
IΤ
     7440-18-8DP, Ruthenium, phosphate complexes
     RL: PREP (Preparation)
        (recovery of, from combustion of simulated spent fuel reprocessing
        solvent contq. ruthenium nitrosyl nitrate, radioactive waste
        disposal in relation to)
     13454-71-2P, Cerium monophosphate
ΙT
```

RL: PREP (Preparation) (recovery of, from combustion of spent TBP solvent contg. cerium nitrate, fuel reprocessing waste issues in relation to)

13537-10-5P, Europium(III) phosphate IT

RL: PREP (Preparation)

(recovery of, from combustion of spent TBP solvent contg. europium nitrate, fuel reprocessing waste issues in relation to)

13465-57-1P, Samarium(III) phosphate IT

RL: PREP (Preparation)

(recovery of, from combustion of spent TBP solvent contg. samarium nitrate, fuel reprocessing waste issues in relation to)

13565-97-4P, Zirconium pyrophosphate IT

RL: PREP (Preparation)

(recovery of, from combustion of spent TBP solvent contg. zirconyl nitrate, fuel reprocessing waste issues in relation to)

7440-07-5P, Plutonium, preparation ΙT

RL: PREP (Preparation)

(recovery of, from phosphoric acid solns. by copptn. with barium phosphate, spent fuel reprocessing solvent waste in relation

7664-38-2P, Phosphoric acid, preparation ΙT

RL: PREP (Preparation)

(recovery of, from tri-Bu phosphate combustion in simulated spent fuel reprocessing solvent treatment process)

- L14 ANSWER 1 OF 2 CAPLUS COPYRIGHT 2002 ACS
- AN 1998:174931 CAPLUS
- DN 128:234216
- TI Manufacture of supplementary cementitious materials from industrial wastes
- AU Mishulovich, Alex; Bhatty, Javed I.; Abbasi, Hamid A.; Rue, David; Olabin, Vladimir M.; Pioro, Leonard S.
- CS Construction Technology Laboratories, Skokie, IL, USA
- Proc. Int. Conf. Incineration Therm. Treat. Technol. (1996), 297-301 Publisher: University of California, Irvine, Irvine, Calif. CODEN: 65TTAP
- DT Conference
- LA English
- CC 58-1 (Cement, Concrete, and Related Building Materials) Section cross-reference(s): 60
- Supplementary cementitious materials (SCM) were manufd. by melting and AΒ vitrification of specially designed blends of wastes with the addn. of inexpensive natural materials (limestone, sand, shale, etc.). This approach opens an outlet for the rational use of wastes and reduces carbon dioxide emission usually assocd. with prodn. of conventional portland cement. The paper summarizes the results of the bench top phase of formulation and testing of SCMs prepd. from Illinois coal ash with the addn. of inexpensive natural or waste materials, such as limestone or cement kiln dust. Selection of the prospective compns. was based on the anal. of phase equil. in the system CaO-SiO2-Al2O3. Compns. were chosen that melt at temps. <1250.degree.. These compns. were realized by mixing the ingredients in the calcd. proportions. Performance of the produced materials was tested in blended cements and concretes. Blended cements incorporating SCMs are not only competitive in terms of strength but have an addnl. advantage of preventing or greatly reducing deleterious chem. reactions between the cement paste and concrete aggregates. This improves the durability of concrete. Unlike conventional portland cements, the supplementary cementing materials should be produced in the form of glasses to provide the necessary chem. reactivity of the product. Besides, vitrification prevents leaching of the trace elements present in the source materials. Submerged gas combustion was suggested and tested as the process of choice for commercialization of this technol. Pilot testing of the submerged combustion melter begins this year in a 250-kg/h test facility. Further studies are under way to finalize the prodn. process parameters and to investigate the products performance.
- ST supplementary cementitious material industrial waste
- IT Wastes

(industrial; manuf. of supplementary cementitious materials from industrial wastes)

IT Cement (construction material)

Recycling

(manuf. of supplementary cementitious materials from industrial wastes)